

## Stereoselective Oxidative Coupling and Asymmetric Hydride Reduction related to (–)-(S)-10,10'-Dihydroxy-9,9'-biphenanthryl

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Oxidative coupling of 9-phenanthrol (**1**) with a (–)-(R)-1,2-diphenylethylamine-copper(II) complex yielded (–)-(S)-10,10'-dihydroxy-9,9'-biphenanthryl (**2**) (98% optical purity); the chiral aluminium hydride reagent modified by (**2**) was found to exhibit enantioface selectivity towards a variety of prochiral carbonyl compounds.

While investigating novel asymmetric hydride reduction,<sup>1</sup> we found that 10,10'-dihydroxy-9,9'-biphenanthryl (**2**) is an excellent chiral modifier in asymmetric LiAlH<sub>4</sub> reduction. In this communication, we report a stereoselective oxidative coupling<sup>2</sup> of 9-phenanthrol (**1**) to provide the optically active (**2**), and an asymmetric hydride reduction utilizing (**2**) as the chiral modifier.

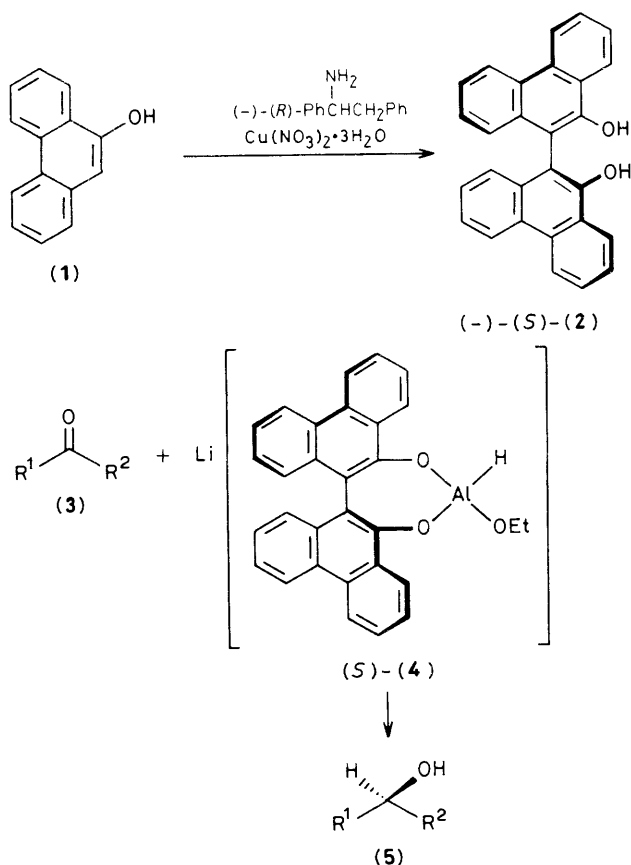
To a solution (–5 °C) of (–)-(R)-1,2-diphenylethylamine {[α]<sub>D</sub><sup>23</sup> –50.8 °C (EtOH),<sup>3</sup> 99% optical purity} (60 mmol) and Cu(NO<sub>3</sub>)<sub>2</sub>·3H<sub>2</sub>O (20 mmol) in methanol (60 ml), a solution of (**1**) (10 mmol) in methanol (20 ml) was added under a nitrogen atmosphere. After stirring at this temperature for 1 h, the reaction mixture was quenched with 2 M HCl and the product

was extracted with diethyl ether. Silica gel chromatography provided 86% (–)-(S)-(**2**),<sup>†</sup> m.p. 234–236 °C, [α]<sub>D</sub><sup>23</sup> –71 °C (CHCl<sub>3</sub>).<sup>4‡</sup>

Following Noyori and co-workers' procedure,<sup>1</sup> a tetrahydrofuran solution of the modified hydride reagent (**4**) was

<sup>†</sup> The (S)-configuration of (–)-(**2**) was determined by chemical correlation with (–)-(S)-2,2'-dihydroxy-3,3'-dimethyl-1,1'-binaphthyl (ref. 5). The optical purity of (–)-(**2**) was determined to be 98% by h.p.l.c. analysis with a column packed with poly(triphenylmethyl methacrylate) (ref. 6).

<sup>‡</sup> (–)-1,2-Diphenylethylamine was recovered (90% yield) without any noticeable loss in optical purity.



prepared *in situ* at room temperature from  $\text{LiAlH}_4$ ,  $(-)-(S)\text{-}(2)$ , and ethanol (1 : 1 : 1 mol ratio), and then cooled to  $-5^\circ\text{C}$ . The prochiral carbonyl compound (3) (1 equiv.) was added to this solution of (4) (3 equiv.) and stirred for 1 h at  $-5^\circ\text{C}$ . Routine work-up involving  $\text{SiO}_2$  chromatography and distillation afforded the results summarized in Table 1. § From these results it can be seen that (i) the chiral hydride reagent (4) exhibits good enantiomeric face selectivity towards the carbonyl compounds having a phenyl group directly linked to the carbonyl centre, (ii) the alcohol products (5) invariably

§ The chiral modifier (2) was recovered without any noticeable loss in yield and optical purity.

**Table 1.** Asymmetric hydride reduction of prochiral carbonyl compounds (3) with  $(S)\text{-}(4)$ .<sup>a</sup>

Substrate ( $\text{R}^1\text{COR}^2$ )	Isolated yield/%	Alcohol (5)	
		$[\alpha]_{\text{D}}^{25}/^\circ$ (c, solvent)	% Enantiomeric excess (configuration)
$\text{R}^1 = \text{Ph}, \text{R}^2 = \text{D}^b$	74	+1.37 (6.25, $\text{C}_5\text{H}_{10}$ )	87 (S) <sup>c</sup>
$\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Me}$	75	-41.8 (2.26, $\text{C}_5\text{H}_{10}$ )	97 (S) <sup>d</sup>
$\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Et}$	78	-44.5 (2.54, $\text{CHCl}_3$ )	98 (S) <sup>e</sup>
$\text{R}^1 = \text{Ph}, \text{R}^2 = \text{CH}_2\text{Ph}$	77	+54.8 (3.52, EtOH)	98 (S) <sup>f</sup>
$\text{R}^1 = \text{CH}_2\text{Ph}, \text{R}^2 = \text{Me}$	76	+13.8 (3.22, $\text{C}_6\text{H}_6$ )	33 (S) <sup>g</sup>
$\text{R}^1 = \text{Bu}^i, \text{R}^2 = \text{Me}$	73	+4.31 (4.26, EtOH)	21 (S) <sup>h</sup>

<sup>a</sup> The reaction was carried out using  $(S)\text{-}(4)$  at  $-5^\circ\text{C}$  for 1 h. <sup>b</sup> 99% Deuteriated (ref. 7). <sup>c</sup> Based on  $[\alpha]_{\text{D}}^{20} + 1.58^\circ$  (c 7.07,  $\text{C}_5\text{H}_{10}$ ) (ref. 8). <sup>d</sup> Based on  $[\alpha]_{\text{D}}^{21} - 43.1^\circ$  ( $\text{C}_5\text{H}_{10}$ ) (ref. 7). <sup>e</sup> Based on  $[\alpha]_{\text{D}} - 45.45^\circ$  (c 5.15,  $\text{CHCl}_3$ ) (ref. 9). <sup>f</sup> Based on  $[\alpha]_{\text{D}}^{18} + 55.9^\circ$  (c 1.40, EtOH) (ref. 10). <sup>g</sup> Based on  $[\alpha]_{\text{D}}^{20} + 41.8^\circ$  (c 5.26,  $\text{C}_6\text{H}_6$ ) (ref. 9). <sup>h</sup> Based on  $[\alpha]_{\text{D}}^{20} + 20.5^\circ$  (EtOH) (ref. 9).

have the (S)-configuration, and (iii) rather low selectivities were found with the aliphatic ketones.

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